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volume but liquids do a liquid takes the shape of its container solids don't change shape solids fill the space they are put in liquids only take the shape of the container they are put in grade 3 matter in ncert solutions for class 11 chemistry chapter 3 you will learn about of periodic classification modern periodic law the periodic table nomenclatures of elements electronic configurations of elements types of elements periodic trends in properties of elements dalton's atomic theory 1 all matter is composed of atoms 2 atoms of a given element are identical atoms of different elements differ from each other 3 atoms cannot be subdivided created or destroyed 4 atoms combine in simple whole number ratios to form compounds 5 in chemical reactions elements are combined separated or the two masses have the same numerical value but the units are different the molecular mass is the mass of 1 molecule while the molar mass is the mass of 6 022 10²³ molecules 13 a 256 528 g mol b 72 150 g mol c 378 103 g mol d 58 080 g mol e 180 158 g mol 15 chapter 3 the structure of matter and the chemical elements 75 3 1 solids liquids and gases 3 2 the chemical elements 3 3 the periodic table of the elements 3 4 the structure of the elements 3 5 common elements 3 6 relating mass to number of particles one doesn't discover new lands without consenting to lose sight of dalton's atomic theory 1 all matter is composed of atoms 2 atoms of a given element are identical in mass size and other proportions and different elements differ 3 atoms cannot be subdivided created or destroyed 4 atoms of different elements combine in simple whole number ratios to form chemical reactions 5 in chemical reactions the process whereby fluids pass through a filter or a filtering medium distillation the process of purifying a liquid by boiling it and condensing its vapors crystallization a separation technique that results in the formation of pure solid particles of a substance from a solution containing the dissolved substance a the combustion of gasoline b the tarnishing of a copper statue c the corrosive action of acid rain on granite d the odor of spearmint gum e none of the above d all of the following can be considered physical properties except a color b boiling point c flammability classification of elements and periodicity in properties class 11 notes firstly in this chapter you will study the development of the periodic table also you will learn the periodic law furthermore the modern periodic table arranges the elements on the basis of their atomic numbers n horizontal rows and eighteen vertical columns chapter 3 atoms and molecules ancient indian and greek philosophers have always wondered about the unknown and unseen form of matter the idea of divisibility of matter was considered long back in india around 500 bc here you can read chapter 3 of class 11 chemistry ncert book also after the chapter you can get links to class 11 chemistry notes ncert solutions important question practice papers etc scroll down for classification of elements and periodicity from ncert book class 11 chemistry book important study material the bohr model gives the same expression for the energy as the quantum mechanical expression and hence both properly account for hydrogen's discrete spectrum an example of getting the right answers for the wrong reasons something that many chemistry students can sympathize with but gives the wrong expression for the angular momentum

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